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### Key indicators

Single-crystal X-ray study T = 293 K Mean  $\sigma$ (C–C) = 0.006 Å R factor = 0.038 wR factor = 0.127 Data-to-parameter ratio = 11.6

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

N-(4-Chlorophenyl)benzenesulfonamide

The crystal structure of the title compound,  $C_{12}H_{10}CINO_2S$ , has been determined. The molecules of the substance form chains with adjacent molecules by means of hydrogen bonds, which create infinite helicoids along the *b* axis. The hydrogenbond network can be described by the graph-set as *C*(4) (infinite chain with four atoms in the repeat pattern).

### Comment

Sulfanyls and sulfonamides are drugs used for the treatment of infections, some fungi and certain protozoa. Other therapeutic applications of the compounds are as diuretic and hypoglycaemic agents. On the other hand, the compounds are very interesting from a fundamental point of view, for studying the relationship between van der Waals interactions and hydrogen-bond topology forming a crystal structure architecture.



A view of the *N*-(4-chlorophenyl)benzenesulfonamide molecule, (I), with the atomic numbering is presented in Fig. 1. The conformational state of the molecule in the crystal structure can be characterized in the following way. The torsion angle O1-S-C1-C2, which characterizes the orientation of the SO<sub>2</sub> group and the phenyl ring Ph1 [C1-C6], is 30.7 (4) Å. The benzene rings are rotated relative to each other by 54.39 (15)°. The torsion angle N1-S-C1-C2, which describes the position of the NH group relative to the Ph1 ring, is -83.4 (3)°, whereas the torsion angle S-N1-C7-C12, which characterizes the location of the SO<sub>2</sub> group with respect to the Ph2 [C7-C12] ring, is -71.1 (4)°.



© 2006 International Union of Crystallography Printed in Great Britain – all rights reserved A view of (I), with the atomic numbering scheme. Displacement ellipsoids are drawn at the 20% probability level.

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### Figure 2

Projection of the molecular packing of (I) along the *a* axis.





The values of the hydrogen-bond geometric parameters are summarized in Table 1. The molecular packing architecture is shown in Figs. 2 and 3. The molecules of (I) form chains with adjacent molecules by means of hydrogen bonds, which create infinite helicoids along the b axis. The hydrogen-bond network can be described by the graph-set assignment introduced by Etter (1990) as C(4) (infinite chain with four atoms in the repeat pattern). The chains form a complex layer structure which can be characterized as follows. The layer includes two types of chains which are situated in such a way that the periphery consists of the chlorophenyl groups, whereas the inner part consists of the unsubstituted phenyl rings. The chlorophenyl fragments of adjacent layers are parallel and interact with each other only by van der Waals forces. The unsubstituted phenyl rings also interact with each other by van der Waals forces, but because of the non-parallel arrangement the energy of these interactions is lower than the analogous one for the chlorophenyl fragments. These conclusions are confirmed by our data from sublimation experiments, which will be published in the near future.

## **Experimental**

The chemical synthesis of the title compound was performed by analogy to procedures described previously (Crosley et al., 1940; Anderson et al., 1942; Gutsche et al., 1974), by reaction of a substituted aromatic amine (here chloroaniline) with benzenesulfonyl

Mo Ka radiation

reflections

T = 293 (2) K

 $h = 0 \rightarrow 12$ 

 $k = 0 \rightarrow 10$  $l = -26 \rightarrow 0$ 

3 standard reflections

frequency: 120 min

intensity decay: 2%

Prism, colourless  $0.5\,\times\,0.4\,\times\,0.1$  mm

 $\theta = 5 - 12^{\circ}$  $\mu = 0.46 \text{ mm}^{-1}$ 

Cell parameters from 35

Crystal data

C12H10CINO2S  $M_r = 267.72$ Orthorhombic, Pbca a = 10.840 (2) Å b = 9.740(1) Å c = 23.596 (3) Å V = 2491.3 (6) Å<sup>3</sup> Z = 8 $D_x = 1.428 \text{ Mg m}^{-3}$ Data collection Enraf-Nonius CAD-4 diffractometer

 $\omega$ -2 $\theta$  scans 1838 measured reflections 1838 independent reflections 1568 reflections with  $I > 2\sigma(I)$  $\theta_{\rm max} = 23.5^\circ$ 

# Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.075P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.038$	+ 2.965P]
$wR(F^2) = 0.127$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.15	$(\Delta/\sigma)_{\rm max} = 0.001$
1838 reflections	$\Delta \rho_{\rm max} = 0.41 \ {\rm e} \ {\rm \AA}^{-3}$
158 parameters	$\Delta \rho_{\rm min} = -0.34 \text{ e } \text{\AA}^{-3}$
H atoms treated by a mixture of	Extinction correction: SHELXL97
independent and constrained	Extinction coefficient: 0.0094 (9)
refinement	

### Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdots A$ $D - H$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$N1 - H1 \cdots O2^{i}$ 0.78 (3)	2.212 (34)	2.993 (4)	175 (4)

Symmetry code: (i)  $-x + \frac{1}{2}, y - \frac{1}{2}, z$ .

C-bound H atoms were positioned geometrically and refined as riding, with C-H = 0.93 Å. Coordinates of the N-bound H atom were determined by an optimization procedure (using a riding model), where  $U_{iso}(H)$  values were set at  $1.2U_{eq}(C,N)$ .

Data collection: CAD-4-PC Software (Enraf-Nonius, 1989); cell refinement: CELDIM in CAD-4-PC Software; data reduction: XCAD4 (Harms & Wocadlo, 1995); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997a); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997a); molecular graphics: XP in SHELXTL (Sheldrick, 1997b); software used to prepare material for publication: CIFTAB in SHELXTL.

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